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Chemistry 25.2 Nuclear
Transformations.

STUDY. PLAY. Nuclear
Force. Attractive force
that acts between all
nuclear particles that
are extremely close
together. Positron.

Particle with the mass
of an electron but a
positive charge. Half
Life. The time required

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Nuclear Transformations

From one-half of the nuclei in a radioisotope sample to decay to products.

Chemistry 25.2

Nuclear Transformations

Flashcards | Quizlet

Chapter 25.2-Nuclear Transformations vocabulary and key concepts. STUDY. PLAY. Nuclear force. An attractive force that acts between all nuclear particles that

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are extremely close together, such as protons and neutrons in a nucleus. Band of stability. The location of stable nuclei on a neutron-vs.-proton plot.

Chapter 25.2-Nuclear Transformations vocabulary and key

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Section 25.2 Nuclear
Transformations 803

Number of Neutrons

vs. Number of Protons

for Stable Nuclei 20 40

60 80 100 Number of

protons 120 20 0 40 60

80 100 Number of

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neutrons 120 140

Neutrons 1 Protons 1

(Ratio 1.0) Ratio 1.5

Ratio 1.3 Band of
stability Ratio 1.2

Stable Nuclei 25.2

Nuclear

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Weather stripping and
insula-

25.2 Nuclear

Transformations 25

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Section Review

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Section Review

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Section Review

Answers

Nuclear transformation is what happens to unstable atom nuclei in while striving to reach stability. The nucleus of any element atom should have specific neutron/proton ratio to be a stable nucleus.

What are examples of nuclear

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Answers

25.2 Lecture 25.2-
Nuclear

Transformations

Radon-222 is a radioactive isotope that is present naturally in the soil in some areas. It has a constant rate of decay. You will learn about decay rates of radioactive substances.

3. The nucleus is held together by the strong nuclear force. The

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strong nuclear force is
the strongest force
known Protons ...

Lecture 25.2-

Nuclear

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Transformations Mary

Beth Smith. Nuclear

chemistry Kris Ann

Ferrer. Ppt nuclear

stability Sherzod

Eshnazarov. 5 nuclear

stability and

radioactive decay

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Sample Problem 25.1 -

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work step by step

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written by community members like you.

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Chapter 25 - Nuclear Chemistry - 25.2

Nuclear ...

25.2 Nuclear

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Stability and Decay
Some nuclei are
unstable ...

Chapter 25

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Chapter 25 - Nuclear Chemistry - 25.2

Nuclear ...

$6/3\text{Li} + 1/0\text{n} \rightarrow \underline{\quad} +$
 $4/2\text{He}$ 1. $6/2\text{He}$ 2. $2/1$
H 3. $3/1\text{H}$ 4. $0/+1\text{Beta}$
5. $5/1\text{H}$

**predict the other
product of the
following nuclear ...**

Answers to Ch. 25

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Section Review

Problems Section

Review 25.1 Part A

Completion 1.

radioactive 2.

radioisotopes 3. nuclei

4. Stable 5. energy 6.

beta 7. Alpha 8. Helium

9. electrons 10. metal

foil 11. Gamma 12.

mass 13. Lead 14.

concrete 15. stop Part

B True-False 16. ST 17.

NT 18. AT 19. NT 20.

AT Part C Matching 21.

b 22. a 23. c 24. e 25.

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ch.25 section review

Answers - Answers

to Ch 25 Section ...

Nuclear energy covers technology of reactors, different types of reactors, their history and development, exploitation in different countries, operating experiences, problems of nuclear waste.

**Answers about
Nuclear Energy**

At this station,
students will be

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watching a short video explaining energy transformations. Students will then answer questions related to the video and record their answers on their lab station sheet. For example: List 6 different types of energy that are found in the video. List 4 different energy transformation examples from the video.

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ENERGY
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LESSON PLAN - A
COMPLETE SCIENCE

...

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Transmutation, conversion of one chemical element into another. A transmutation entails a change in the structure of atomic nuclei and hence may be induced by a nuclear reaction (q.v.), such as neutron capture, or occur spontaneously by radioactive decay, such as alpha decay and beta decay (qq.v.).

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of base metals (such as mercury, tin, copper, lead) into precious metals (gold ...

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