

Define Solution Chemistry Terms

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Define Solution Chemistry Terms

A solution consists of a solute and a solvent.The solute is the substance that is dissolved in the solvent. The amount of solute that can be dissolved in solvent is called its solubility.For example, in a saline solution, salt is the solute dissolved in water as the solvent.

Solution Definition in Chemistry - ThoughtCo

Solution (chemistry) synonyms, Solution (chemistry) pronunciation, Solution (chemistry) translation, English dictionary definition of Solution (chemistry). n. 1. a. A method or process of dealing with a problem: sought a solution to falling enrollments. b.

Solution (chemistry) - definition of Solution (chemistry) ...

Solution, in chemistry, a homogenous mixture of two or more substances in relative amounts that can be varied continuously up to what is called the limit of solubility. The term solution is commonly applied to the liquid state of matter, but solutions of gases and solids are possible.

solution | Definition & Examples | Britannica

In chemistry, a solution is a special type of homogeneous mixture composed of two or more substances. In such a mixture, a solute is a substance dissolved in another substance, known as a solvent.The mixing process of a solution happens at a scale where the effects of chemical polarity are involved, resulting in interactions that are specific to solvation.

Solution - Wikipedia

Solution - A solution is a mixture formed when a solid, liquid or gaseous substance is homogeneously mixed with a liquid. Likewise, a solvent is a substance in which another substance dissolves.

Solution - Definition, Properties, Types, Videos & Examples

Similarly, a solution of known concentration has been standardized. To prepare a standard solution, a known mass of solute is dissolved and the solution is diluted to a precise volume. Standard solution concentration is usually expressed in terms of molarity (M) or moles per liter (mol/L).

Standard Solution Definition - Chemistry Glossary

a solution that contains less solute than a saturated solution under the existing conditions. ... Chemistry -- Terms and Definitions. 87 terms. fmmlee. Unit 4 Vocab: the federal courts. 28 terms. hfaustlittle. Hydrostatic Pressure VS Osmotic Pressure. 14 terms. krconatser87. Chapter 5 - Solutions.

Chapter 13 (Solutions) Definitions Flashcards | Quizlet

A molar solution is defined as an aqueous solution that contains 1 mole (gram-molecular weight) of a compound dissolved in 1 liter of a solution. In other words, the solution has a concentration of 1 mol/L or a molarity of 1 (1M).

What is a Molar Solution? - Definition from Corrosionpedia

Supersaturated Solution - A supersaturated solution is one in which more solute is dissolved than is necessary to make a saturated solution. A supersaturated solution is unstable solute molecules may crash out of solution given the slightest perturbation. Learn more about supersaturated solutions at BYJU'S.

Supersaturated Solution - Definition, Examples ...

This glossary of chemistry terms is a list of terms and definitions relevant to chemistry, including chemical laws, diagrams and formulae, laboratory tools, glassware, and equipment.Chemistry is a physical science concerned with the composition, structure, and properties of matter, as well as the changes it undergoes during chemical reactions; it features an extensive vocabulary and a ...

Glossary of chemistry terms - Wikipedia

Solutions in science. As previously learned, solutions are homogeneous, or equal throughout, and contain a solute and solvent. As a reminder, the solute is present in less amount than the solvent.

What is a Solution in Science? - Definition & Examples ...

Search the Dictionary for More Terms. Return to top of page. Copyright © 2020 chemicool.com Chemistry Dictionary | Birth of the Elements | Tools | Periodic Table ...

Definition of solution - Chemistry Dictionary

solution [so-loo' shun] 1. a homogeneous mixture of one or more substances (solute) dispersed molecularly in a sufficient quantity of dissolving medium (solvent). 2. in pharmacology, a liquid preparation of one or more soluble chemical substances, which are usually dissolved in water. For names of specific solutions, see under the name. 3. the process ...

Solution (chemistry) | definition of Solution (chemistry) ...

solution, in chemistry, homogeneous mixture mixture, in chemistry, a physical combination of two or more pure substances (i.e., elements or compounds). A mixture is distinguished from a compound, which is formed by the chemical combination of two or more pure substances in a fixed, definite proportion.

Solution (chemistry) | Article about Solution (chemistry) ...

Solution definition, the act of solving a problem, question, etc.: The situation is approaching solution. See more.

Solution | Definition of Solution at Dictionary.com

To start browsing the dictionary, choose a letter to list all the terms starting with that letter. You can also search our dictionary using the search box provided. Our collection of chemistry terms and definitions cover all areas, including: general chemistry science, organic chemistry, inorganic, biological, physical, environmental and analytical chemistry.

Chemistry Dictionary, Terms & Definitions - Chemistry ...

The concentration of a solution is the proportion of the solute to solvent. If there is a lot of solute in a solution, then it is "concentrated". If there is a low amount of solute, then the solution is said to be "diluted." Miscible and immiscible When two liquids can be mixed to form a solution they are called "miscible."

Chemistry for Kids: Solutions and Dissolving

Aqueous is a term used to define a system that involves water. The word aqueous is also applicable to describe a solution or mixture in which water is the solvent. A substance will form an aqueous solution or not, it depends on the nature of its chemical bonds.