

Properties Of Metals Yahoo Answers

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Properties Of Metals Yahoo Answers

Of primary concern in aircraft maintenance are such general properties of metals and their alloys as hardness, brittleness, malleability, ductility, elasticity, toughness, density, fusibility,...

properties of metals? | Yahoo Answers

Iron is the main component of steel, and it is used in the production of alloys or solid solutions of various metals, as well as some non-metals, particularly carbon. The many iron-carbon alloys, which have very different properties, are discussed in the article on steel.

Properties of metals? | Yahoo Answers

Traditionally, metals have certain characteristic physical properties: they are usually shiny (they have metallic luster), have a high density, are ductile and malleable, usually have a high...

what are the properties of metal? | Yahoo Answers

Noble metals include copper, palladium, silver, platinum, and gold. 2. Alkali Metals are very reactive. They have low melting points and are soft enough to be cut with a knife. Potassium and sodium...

5 properties of metals? | Yahoo Answers

1. metals are ductile and malleable i.e they can be made into wires and beaten into sheets 2. metals have high melting and boiling points with the notable exceptions of gallium and mercury both of...

HELP!!!!!!! What are the four main properties of metals ...

Metals are shiny solids are room temperature (except mercury), with characteristic high melting points and densities. Many of the properties of metals, including large atomic radius, low ionization...

what are metals and what are the properties of metals ...

For the best answers, search on this site <https://shorturl.im/5xNAh>. properties: 1. they conduct heat 2. they conduct electricity 3. they are solid at room temperature (except mercury) 4. they are shiny 5. are malleable and ductile 6 different metals: 1.

What are the 5 properties of metals?!? | Yahoo Answers

Properties of metals : Metals are solid at room temp. They are generally hard and strong. They have metallic lustre.

properties of metals and non-metals? | Yahoo Answers

Properties of Metals: * Metals are usually opaque, hard solids with high density and have shiny, silvery luster, when cut or scratched. * Metals have strong metallic bonds that make them malleable, ductile and easy to bend. They have high melting points, boiling points and density. Metals can be hammered into new shapes because they are malleable.

what r 4 examples of properties of ... - Yahoo Answers

Physical Properties : The metals of Group 2 are harder and denser than sodium and potassium, and have higher melting points. These properties are due largely to the presence of two valence...

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what are the properties of non-metals? Answer Save. 1 Answer. Relevance. Princess.Drama. 1 decade ago. Favorite Answer. dull colour, not shiny, non malleable,:) 0 0. Still have questions? Get your answers by asking now. Ask Question + 100. Join Yahoo Answers and get 100 points today. Join. Trending Questions. Trending Questions. At ...

what are the properties of non-metals? | Yahoo Answers

Metals are ductile and malleable, so their shape can be easily changed into thin wires or sheets. Metals will corrode, gradually wearing away like rusting iron. Nonmetals, on the right side of the periodic table, are very different from metals. Their surface is dull and they are poor conductor of heat and electricity.

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what effects does recrystallization have on properties of metals? Answer Save. 2 Answers. Relevance. shadouse. Lv 6. 1 decade ago. Favourite answer. It often weakens them as it creates fractures. 0 0. 1233345. 1 decade ago. sedrfqtyhujik. 0 0. Still have questions? Get answers by asking now. Ask question + 100. Join Yahoo Answers and ...

what effects does recrystallization have on properties of ...

5 answers Physics help! ? Acceleration due to gravity of $g = 6.794 \text{ m/s}^2$ and want to jump safely into the water below after leaping from a 9.00 meter high-cliff. ?

what effects does recrystallization have ... - Yahoo Answers

Lesson revising the physical properties of metals. This website and its content is subject to our Terms and Conditions.

Properties of Metals | Teaching Resources

Chemistry, 15.10.2020 03:07, Sowdamini Describe some typical properties of alkali metals and noble gases

Describe some typical properties of alkali metals and ...

Properties of metals are as follows: Metals are great conductors of electricity. Brittle is a characteristic of nonmetals, malleable, ductile and good conductor are characteristics of metals.

Properties of metals and non-metals? - Answers

Answers: 2 on a question: Elements in which the d-sublevel is being filled have the properties of * O metals. O nonmetals. O metalloids. O gases

Elements in which the d-sublevel is being filled have the ...

By considering the structure and bonding of metals which you have studied, we can explain these properties. Metals are strong, with high melting and boiling points. Metals have high melting and boiling points as a result of the strong electrostatic attraction between the positively charged metal ions (we call positively charged ions cations), and sea of delocalised electrons.

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